

# Preparation, Characterization and Crystal Structure of the Room Temperature Phase of $[(\text{CH}_3)(\text{C}_6\text{H}_5)_3\text{P}]_2[\text{ZnBr}_4]$ : A Member of the $\text{A}_2\text{BX}_4$ Family

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The compound bis(methyltriphenylphosphonium) tetrabromozincate(II),  $[\text{C}_{19}\text{H}_{18}\text{P}]_2[\text{ZnBr}_4]$ ,  $M_r = 939.640$ , has a monoclinic unit cell, space group  $P2_1$ . The lattice parameters are  $a = 9.7693(4) \text{ \AA}$ ,  $b = 12.5508(4) \text{ \AA}$ ,  $c = 16.5372(6) \text{ \AA}$ ,  $\alpha = 90.00^\circ$ ,  $\beta = 105.2670(11)^\circ$ ,  $\gamma = 90.00^\circ$ ,  $V = 1956.11(11) \text{ \AA}^3$ ,  $Z = 2$ ,  $D_x = 1.595 \text{ mg m}^{-3}$  at  $T = 298 \text{ K}$ . The structure consists of one distorted  $[\text{ZnBr}_4]^{2-}$  tetrahedron and two  $[(\text{CH}_3)(\text{C}_6\text{H}_5)_3\text{P}]^+$  cations. Differential scanning calorimetry indicates a continuous second-order transition at  $(276 \pm 2) \text{ K}$  that may be classified as a commensurate to incommensurate transformation. A first-order transition to a higher symmetry is associated with a four-fold rotation of the  $[\text{ZnBr}_4]^{2-}$  ion and a change of entropy  $\Delta S = 22.92 \text{ J/(K} \cdot \text{mol)}$  at  $T = (362 \pm 3) \text{ K}$ .

Dilatometric measurements showed a decrease of the lattice parameters in the temperature range 230 – 260 K, confirmed the transition at  $(276 \pm 2) \text{ K}$ , and indicated the presence of a third transition at 282 K.

*Key words:* Crystal Structure; Phase Transition; Commensurate-Incommensurate.

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